

Stereoconvergent Palladium-Catalyzed Carbonylation of Both *E* and *Z* Isomers of a 2-Trifloxy-1,3-butadiene

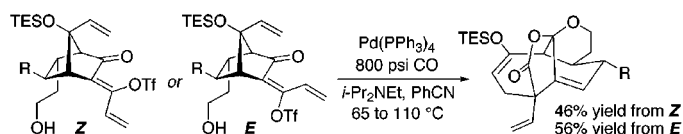
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ABSTRACT

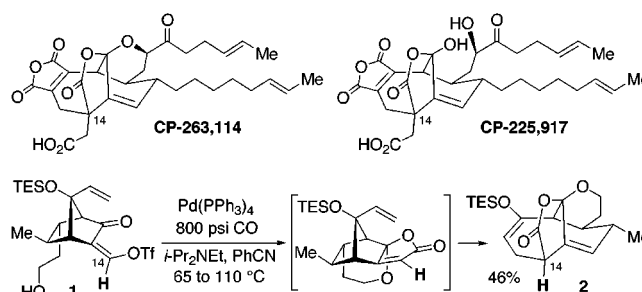


Carbonylation of the illustrated *Z*-tetrasubstituted enol triflate followed by tandem silyloxy-Cope rearrangement leads to the CP-263,114 core ring system with the all-carbon quaternary stereocenter intact in 46% yield. Subjection of the corresponding *E* isomer to the same conditions gives the same product in 56% yield. This observation is explained by a mechanism involving isomerization of a π -allyl palladium species involving an allenic intermediate.

CP-263,114 and CP-225,917¹ have, due primarily to their unique and challenging molecular architecture, attracted the attention of synthetic chemists. In addition to many diverse strategies,² total syntheses have been reported by Nicolaou,³ Danishefsky,⁴ and Shair.⁵ Our own approach revolves around a late-stage tandem carbonylation-Cope rearrangement that delivers the entire core ring system of the CP structures (1

→ 2, Scheme 1).⁶ One of the many challenges posed by these targets—and one which was unaddressed in our model

Scheme 1

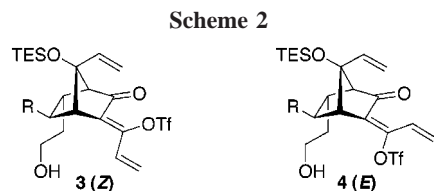


study—is the all carbon quaternary stereocenter at C(14) (highlighted proton, Scheme 1).⁷ Herein we describe our efforts to extend our approach to incorporate an appropriate substituent at C(14), and an unusual observation regarding the chemistry of 2-trifloxy-1,3-butadienes.

- (1) (a) Dabrah, T. T.; Harwood, H. J.; Huang, L. H.; Jankovich, N. D.; Kaneko, T.; Li, J.-C.; Lindsey, S.; Moshier, P. M.; Subashi, T. A.; Therrien, M.; Watts, P. C. *J. Antibiot.* **1997**, *50*, 1–7. (b) Dabrah, T. T.; Kaneko, T.; Massefski, W. Jr.; Whipple, E. B. *J. Am. Chem. Soc.* **1997**, *119*, 1594–1598.
(2) (a) Davies, H. M. L.; Calvo, R.; Ahmed, G. *Tetrahedron Lett.* **1997**, *38*, 1737–1740. (b) Nicolaou, K. C.; Härter, M. W.; Boulton, L.; Jandeleit, B. *Angew. Chem., Int. Ed. Engl.* **1997**, *36*, 1194–1196. (c) Nicolaou, K. C.; Postema, M. H. D.; Miller, N. D.; Yang, G. *Angew. Chem., Int. Ed. Engl.* **1997**, *36*, 2821–2823. (d) Sgarbi, P. W. M.; Clive, D. L. J. *Chem. Commun.* **1997**, 2157–2158. (e) Armstrong, A.; Critchley, T. J.; Mortlock, A. A. *Synlett* **1998**, 552–553. (f) Kwon, O.; Su, D.-S.; Meng, D.; Deng, W.; D'Amico, D. C.; Danishefsky, S. J. *Angew. Chem., Int. Ed. Engl.* **1998**, *37*, 1877–1880. (g) Kwon, O.; Su, D.-S.; Meng, D.; Deng, W.; D'Amico, D. C.; Danishefsky, S. J. *Angew. Chem., Int. Ed. Engl.* **1998**, *37*, 1880–1882. (h) Waizumi, N.; Itoh, T.; Fukuyama, T. *Tetrahedron Lett.* **1998**, *39*, 6015–6018. (i) Chen, C.; Layton, M. E.; Shair, M. D. *J. Am. Chem. Soc.* **1998**, *120*, 10784–10785. (j) Nicolaou, K. C.; Baran, P. S.; Jautelat, R.; He, Y.; Fong, K. C.; Choi, H. S.; Yoon, W. H.; Zhong, Y.-L. *Angew. Chem., Int. Ed. Engl.* **1999**, *38*, 549–552. (k) Clive, D. L. J.; Sun, S.; He, X.; Zhang, J.; Gagliardini, V. *Tetrahedron Lett.* **1999**, *40*, 4605–4609. (l) Clive, D. L. J.; Zhang, J. *Tetrahedron* **1999**, *55*, 12059–12068. (m) Yoshimitsu, T.; Yanagiya, M.; Nagaoka, H. *Tetrahedron Lett.* **1999**, *40*, 5215–5218. (n) Sulikowski, G. A.; Agnelli, F.; Corbett, R. M.; *J. Org. Chem.* **2000**, *65*,

- 337–342. (o) Crimmins, M. T.; Hauser, E. B. *Org. Lett.* **2000**, *2*, 281–284. For recent reviews, see: (p) Diederichsen, U. *Nachr. Chem. Technol. Lab.* **1999**, *47*, 1423–1427. (q) Hepworth, D. *Chem. Ind. (London)* **2000**, *2*, 59. (r) Starr, J. T.; Carreira, E. M. *Angew. Chem., Int. Ed.* **2000**, *39*, 1415–1421.

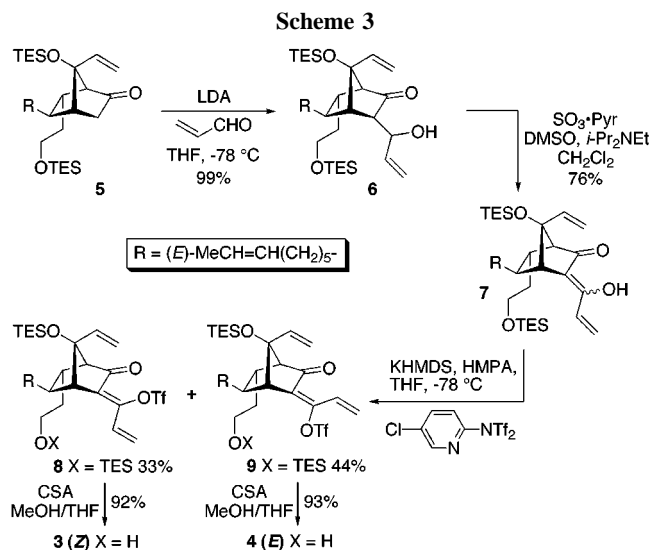
After considering several options, we targeted a vinyl group for the C(14) substituent because it was deemed unlikely to interfere with the carbonylation-Cope sequence, and due to its synthetic equivalence to the requisite acetic acid. This decision seemingly necessitated the stereoselective synthesis of (*Z*)-tetrasubstituted enol triflate **3** (Scheme 2).



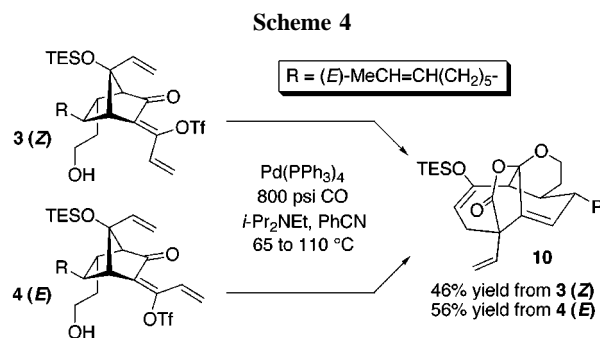
We were intrigued, however, by the possibility that the (*E*)-enol triflate **4** might also serve as an effective precursor to the desired carbonylation-Cope rearrangement product. If so, the strategic advantage would be substantial in that an (*E*)-selective or even a nonselective synthesis of the enol triflates **3** and **4** would be equally useful.

Our studies commenced with ketone **5**.⁸ Attempted Claisen condensation with acrylate esters under a variety of conditions proved fruitless, and as a result an aldol-oxidation sequence was investigated. Treatment of ketone **5** with lithium diisopropylamide (LDA) and treatment of the resultant enolate with acrolein in THF at $-78\text{ }^{\circ}\text{C}$ led to a nearly quantitative yield of aldol **6** as a single diastereomer (stereochemistry unassigned). Oxidation according to the Parikh-Doering protocol⁹ then delivered β -diketone **7** in 76% yield. The critical enol triflate formation was then investigated under a variety of conditions.¹⁰ While the regioselectivity for the exocyclic enol triflate was consistently $>10:1$ (controlled by the strain in the endocyclic enolate), the stereoselectivity of exocyclic enol triflate formation was found to vary widely with conditions. Enolization of **7** with potassium bis(trimethylsilyl)amide (KHMDs) in the presence of hexamethylphosphoramide (HMPA) in THF at $-78\text{ }^{\circ}\text{C}$, and treatment of the resulting anion with Comins's triflating reagent¹¹ gave a 1:1.3 mixture of *Z*:*E*-enol triflates **8** and **9** in 77% yield. Conditions that resulted in a *Z* selective

($>10:1$) reaction were identified (KHMDs, TF_2O , Et_2O , $-78\text{ }^{\circ}\text{C}$), however the yield of **3** (*Z*) was an unacceptable 44%. In preparation for the carbonylation-Cope rearrangement, the primary triethylsilyl (TES) ethers of **8** and **9** were hydrolyzed with catalytic camphorsulfonic acid (CSA) in MeOH/THF giving **3** (*Z*) and **4** (*E*) in 92% and 93% yields, respectively.



We first examined the carbonylation-Cope rearrangement of enol triflate **3** to confirm that the C(14) vinyl group would not interfere with the reaction. Indeed, subsection of **3** to the carbonylation conditions developed previously resulted in the isolation of the CP-core fragment **10** in 46% yield (Scheme 4). We then turned our attention to the “wrong”



E-enol triflate **4**. Remarkably, subsection of **4** to the same carbonylation conditions led to the isolation of **10** in 56% yield.

Mechanistically, we propose the scenario depicted in Scheme 5. Following insertion of Pd(0) into the *E*-enol triflate **4**, CO migratory insertion would be expected to be facile. However, this should be a mechanistic dead-end in that the only potential trap for the palladium-acyl is an *intermolecular* reaction with the primary alcohol. The high dilution of the reaction is likely helpful in this regard. Thus,

(3) (a) Nicolaou, K. C.; Baran, P. S.; Zhong, Y.-L.; Choi, H. S.; Yoon, W. H.; He, Y.; Fong, K. C. *Angew. Chem., Int. Ed. Engl.* **1999**, *38*, 1669–1675. (b) Nicolaou, K. C.; Baran, P. S.; Zhong, Y.-L.; Fong, K. C.; He, Y.; Yoon, W. H.; Choi, H. S. *Angew. Chem., Int. Ed. Engl.* **1999**, *38*, 1676–1678. (c) Nicolaou, K. C.; Jung, J.-K.; Yoon, W. H.; He, Y.; Zhong, Y.-L.; Baran, P. S. *Angew. Chem., Int. Ed.* **2000**, *39*, 1829–1832.

(4) Meng, D.; Qiang, T.; Danishefsky, S. J. *Angew. Chem., Int. Ed.* **1999**, *38*, 3197–3201.

(5) Chen, C.; Layton, M. E.; Sheehan, S. M.; Shair, M. D. *J. Am. Chem. Soc.* **2000**, *122*, 7424.

(6) Bio, M. M.; Leighton, J. L. *J. Am. Chem. Soc.* **1999**, *121*, 890–891.

(7) Clive has addressed the construction of the C(14) quaternary stereocenter with a Cope rearrangement strategy. See ref 2k,l.

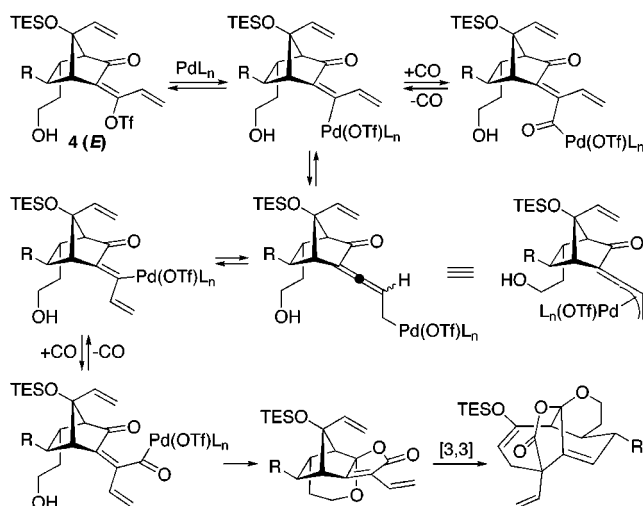
(8) The synthesis of ketone **5** is straightforward and proceeds in 11 steps from commercially available materials. Details will be reported elsewhere.

(9) Parikh, J. R.; Doering, W. von E. *J. Am. Chem. Soc.* **1967**, *89*, 5505–5507.

(10) For a review on the preparation and uses of enol triflates, see: Scott, W. J.; McMurry, J. E. *Acc. Chem. Res.* **1988**, *21*, 47–54.

(11) Comins, D. L.; Dehghani, A. *Tetrahedron Lett.* **1992**, *33*, 6299–6302.

Scheme 5



the initially formed *E*-vinylpalladium species may be expected to isomerize via the indicated π -allyl palladium complex to the *Z*-vinylpalladium complex. From there, CO insertion and trapping of the resultant palladium-acyl with the hemiketal formed from the ketone and primary alcohol lead to the illustrated unsaturated lactone, which then rearranges to give the observed product.

The generation of allenic π -allyl palladium complexes of the type proposed here from buta-2,3-dien-1-ols and their derived acetates, carbonates and phosphates is well-precedented.¹² Various traps have been used (alkyl Zn and Mg, sodium ethylmalonate, CO/MeOH, CO/H₂O, organoboranes) and with the exception of the malonate reaction the products are 2-substituted-1,3-butadienes. 1,3-Diene syntheses involv-

(12) (a) Kleijn, H.; Westmijze, H.; Meijer, J.; Vermeer, P. *Recl. Trav. Chim. Pays-Bas* **1983**, *102*, 378–380. (b) Djahanbini, D.; Cazes, B.; Gore, J. *Tetrahedron Lett.* **1984**, *25*, 203–206. (c) Nokami, J.; Maihara, A.; Tsuji, J. *Tetrahedron Lett.* **1990**, *31*, 5629–5630. (d) Ni, Z.; Padwa, A. *Synlett* **1992**, 869–870. (e) Piotti, M. E.; Alper, H. *J. Org. Chem.* **1994**, *59*, 1956–

ing similar intermediates but starting from 2-butyne-1,4-diol dicarbonates are known as well.¹³ The present work constitutes the first example of the isomerization of one geometric isomer of a 2-substituted-1,3-butadiene to the other. Synthetically this discovery has important ramifications in that the need for a relatively difficult stereospecific synthesis of the *Z*-enol triflate has been obviated. In essence, a stereospecific synthesis of the C(14) quaternary stereocenter has been “paid for” with a nonselective synthesis of the tetrasubstituted enol triflate. The application of this strategy to a total synthesis of CP-263,114 is in progress.

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Supporting Information Available: Experimental procedures and characterization data for **6–9**, **3**, **4**, and **10**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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1957. (f) Moriya, T.; Furuuchi, T.; Miyaura, N.; Suzuki, A. *Tetrahedron* **1994**, *50*, 7961–7968. (g) Imada, Y.; Vasapollo, G.; Alper, H. *J. Org. Chem.* **1996**, *61*, 7982–7983. (h) Uemura, K.; Inoue, Y. *Appl. Organomet. Chem.* **2000**, *14*, 8–13.

(13) Kiji, J.; Okano, T.; Fujii, E.; Tsuji, J. *Synthesis* **1997**, 869–???. (b) Bohmer, J.; Grigg, R. *Tetrahedron* **1999**, *55*, 13463–13470. See also: (c) Gevorgyan, V.; Kadowaki, C.; Salter, M. M.; Kadota, I.; Shinichi, S.; Yamamoto, Y. *Tetrahedron* **1997**, *53*, 9097–9106.